

Problem 1C.2

The wall collision frequency. It is desired to find the frequency Z with which the molecules in an ideal gas strike a unit area of a wall from one side only. The gas is at rest and at equilibrium with a temperature T and the number density of the molecules is n . All molecules have a mass m . All molecules in the region $x < 0$ with $u_x > 0$ will hit an area S in the yz -plane in a short time Δt if they are in the volume $Su_x\Delta t$. The number of wall collisions per unit area per unit time will be

$$\begin{aligned} Z &= \frac{\int_{-\infty}^{\infty} \int_{-\infty}^{\infty} \int_0^{+\infty} (Su_x\Delta t) f(u_x, u_y, u_z) du_x du_y du_z}{S\Delta t} \\ &= n \left(\frac{m}{2\pi kT} \right)^{3/2} \left(\int_0^{+\infty} u_x \exp(-mu_x^2/2kT) du_x \right) \\ &\quad \left(\int_{-\infty}^{+\infty} \exp(-mu_y^2/2kT) du_y \right) \left(\int_{-\infty}^{+\infty} \exp(-mu_z^2/2kT) du_z \right) \\ &= n \sqrt{\frac{kT}{2\pi m}} = \frac{1}{4} n \bar{u} \end{aligned} \tag{1C.2-1}$$

Verify the above development.

Solution

Suppose we have an xyz -coordinate system immersed in an ideal gas. In order for molecules to strike the yz -plane within a certain time interval Δt , the molecules have to be within a distance of $u_x\Delta t$ to do so, u_x being the x -component of their velocity. If we consider a small area on the yz -plane S , then all the molecules within the volume $Su_x\Delta t$ will collide with this area. The Maxwell–Boltzmann distribution function,

$$f(u_x, u_y, u_z) = n \left(\frac{m}{2\pi kT} \right)^{3/2} \exp\left(-\frac{mu^2}{2kT}\right), \tag{1}$$

allows us to find the number of molecules in an ideal gas at rest between certain velocities.

$f(u_x, u_y, u_z) du_x du_y du_z$ represents the number of molecules per unit volume that has velocity between $\langle u_x, u_y, u_z \rangle$ and $\langle u_x + du_x, u_y + du_y, u_z + du_z \rangle$. So then this quantity times the volume $(Su_x\Delta t)f(u_x, u_y, u_z) du_x du_y du_z$ represents the number of molecules in the volume $Su_x\Delta t$ that has velocity between $\langle u_x, u_y, u_z \rangle$ and $\langle u_x + du_x, u_y + du_y, u_z + du_z \rangle$. Integrating over all possible velocities, we obtain the total number of molecules within the volume that will collide with the surface.

$$\int_{-\infty}^{\infty} \int_{-\infty}^{\infty} \int_0^{\infty} (Su_x\Delta t) f(u_x, u_y, u_z) du_x du_y du_z$$

The molecules are assumed to travel from left to right only, so the integral in du_x only goes from 0 to ∞ . Dividing this by the area S , we get the number of collisions per unit area.

$$\frac{\int_{-\infty}^{\infty} \int_{-\infty}^{\infty} \int_0^{\infty} (Su_x\Delta t) f(u_x, u_y, u_z) du_x du_y du_z}{S}$$

Dividing this by the time interval Δt , we get the number of collisions per unit area per unit time Z .

$$Z = \frac{\int_{-\infty}^{\infty} \int_{-\infty}^{\infty} \int_0^{\infty} (Su_x\Delta t) f(u_x, u_y, u_z) du_x du_y du_z}{S\Delta t}$$

Bring out the constant from the integral and cancel it with the one in the denominator.

$$Z = \frac{\mathcal{S}\Delta t \int_{-\infty}^{\infty} \int_{-\infty}^{\infty} \int_0^{\infty} u_x f(u_x, u_y, u_z) du_x du_y du_z}{\mathcal{S}\Delta t}$$

Substitute the formula for $f(u_x, u_y, u_z)$ in equation (1).

$$Z = \int_{-\infty}^{\infty} \int_{-\infty}^{\infty} \int_0^{\infty} u_x n \left(\frac{m}{2\pi kT} \right)^{3/2} \exp\left(-\frac{mu^2}{2kT}\right) du_x du_y du_z.$$

Bring the constant in front of the integral.

$$Z = n \left(\frac{m}{2\pi kT} \right)^{3/2} \int_{-\infty}^{\infty} \int_{-\infty}^{\infty} \int_0^{\infty} u_x \exp\left(-\frac{mu^2}{2kT}\right) du_x du_y du_z.$$

Let $a = m/2kT$ to make the integral easier to write and substitute $u^2 = u_x^2 + u_y^2 + u_z^2$.

$$Z = n \left(\frac{a}{\pi} \right)^{3/2} \int_{-\infty}^{\infty} \int_{-\infty}^{\infty} \int_0^{\infty} u_x e^{-a(u_x^2 + u_y^2 + u_z^2)} du_x du_y du_z.$$

Split up the triple integral into three single integrals.

$$Z = n \left(\frac{a}{\pi} \right)^{3/2} \left(\int_0^{\infty} u_x e^{-au_x^2} du_x \right) \left(\int_{-\infty}^{\infty} e^{-au_y^2} du_y \right) \left(\int_{-\infty}^{\infty} e^{-au_z^2} du_z \right)$$

The integrals in du_y and du_z are known. Use the substitution,

$$\begin{aligned} v &= au_x^2 \\ dv &= 2au_x du_x \quad \rightarrow \quad \frac{dv}{2a} = u_x du_x, \end{aligned}$$

to solve the first one.

$$Z = n \left(\frac{a}{\pi} \right)^{3/2} \left(\int_0^{\infty} e^{-v} \frac{dv}{2a} \right) \left(\sqrt{\frac{\pi}{a}} \right) \left(\sqrt{\frac{\pi}{a}} \right)$$

Evaluate it.

$$Z = n \left(\frac{a}{\pi} \right)^{3/2} \left(\frac{1}{2a} \right) \left(\sqrt{\frac{\pi}{a}} \right) \left(\sqrt{\frac{\pi}{a}} \right)$$

Simplify the result.

$$Z = \frac{n}{2} \frac{1}{\sqrt{\pi a}}$$

Substitute the expression for a .

$$\begin{aligned} Z &= \frac{n}{2} \sqrt{\frac{2kT}{\pi m}} \\ &= \frac{n}{4} \sqrt{\frac{8kT}{\pi m}} \end{aligned}$$

Therefore,

$$Z = \frac{1}{4} n \bar{u},$$

where \bar{u} is the mean molecular speed.