

## Exercise 2

Assume that the half life of the radium in a piece of lead is 1600 years. How much radium will be lost in 100 years?

### Solution

Let the amount of radium in the lead be  $x$ . The rate that  $x$  decreases with respect to time is proportional to the amount present.

$$\frac{dx}{dt} \propto -x$$

The minus sign is included because the radium decomposes; it doesn't accumulate. Change this proportionality to an equation by introducing a proportionality constant  $k$ .

$$\frac{dx}{dt} = -kx$$

Solve this ODE by separating variables.

$$\frac{dx}{x} = -k dt$$

Integrate both sides.

$$\int \frac{dx}{x} = - \int k dt$$

Evaluate the integrals.

$$\ln |x| = -kt + C$$

The absolute value sign is included since the logarithm argument cannot be negative. Exponentiate both sides.

$$e^{\ln |x|} = e^{-kt+C}$$
$$|x| = e^C e^{-kt}$$

Remove the absolute value sign by placing  $\pm$  on the right side.

$$x(t) = \pm e^C e^{-kt}$$

Use a new arbitrary constant  $A$  for  $\pm e^C$ .

$$x(t) = A e^{-kt}$$

Assuming there's an amount of radium  $x = x_0$  initially at  $t = 0$ , then

$$x(0) = A e^0 = x_0 \quad \rightarrow \quad A = x_0.$$

As a result, the solution to the ODE is

$$x(t) = x_0 e^{-kt}.$$

Use the fact that the half-life of the radium in a piece of lead is 1600 years to determine  $k$ .

$$0.50x_0 = x_0 e^{-k(1600)}$$

Solve for  $k$ .

$$0.5 = e^{-1600k}$$

$$\ln 0.5 = \ln e^{-1600k}$$

$$\ln 0.5 = -1600k \ln e$$

$$k = -\frac{1}{1600} \ln 0.5 \approx 0.000433217$$

The solution to the ODE then becomes

$$x(t) = x_0 \exp \left[ \left( \frac{1}{1600} \ln 0.5 \right) t \right].$$

Therefore, after 100 years,

$$x(100) = x_0 \exp \left[ \left( \frac{1}{1600} \ln 0.5 \right) 100 \right] \approx 0.9576x_0 \Rightarrow 4.24\% \text{ of the radium is lost.}$$